

ISOBHARANGIN, A NEW BIOGENETICALLY SIGNIFICANT DITERPENOID QUINONEMETHIDE  
FROM PYGMACOPREMNA HERBACEA (ROXB.) MOLDENKE

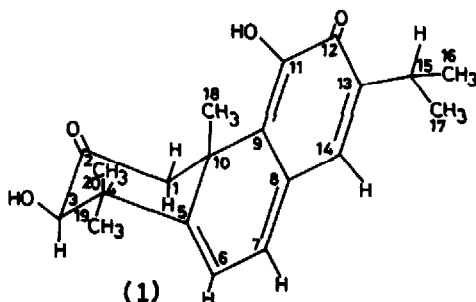
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**Abstract** - Isobharangin isolated from the hexane extract of the root nodules of the Ayurvedic drug, *Pygmacopremna herbacea* (Roxb) Moldenke has been shown to be 3 $\beta$ (R),11-dihydroxy-5,7,9,13-abietatetraene-2,12-dione (1) unequivocally from a consideration of its mass spectrum, UV, IR,  $^1\text{H}$  NMR and CD spectra, long range couplings in its  $^{13}\text{C}$  NMR spectrum and its 2D NOESY spectrum. It represents the first naturally occurring diterpenoid quinonemethide derived from an abietane skeleton with an unusual substitution pattern.

The isolation of three pigments designated bharangin, isobharangin and bharanginin from the hexane extract of the root nodules of the Ayurvedic drug, *Pygmacopremna herbacea* (Roxb) Moldenke syn. *Premna herbacea* (Roxb) (Verbenaceae, Telugu: Gantubharangi) and structure elucidation of bharangin<sup>1</sup> and bharanginin<sup>2</sup> have been reported earlier. Isobharangin, brown solid, m.p.74-76°C (petroleum ether, 40-60°),  $[\alpha]_D^{26} = -465^\circ$  (c, 0.115,  $\text{CHCl}_3$ ) and molecular formula,  $\text{C}_{20}\text{H}_{24}\text{O}_4$  ( $M^+$ , 328.1674) is isomeric with bharangin<sup>1</sup>. Its UV absorption  $\lambda_{\text{max}}$  (Methanol): 418 nm ( $\log \epsilon$  3.64); IR absorptions ( $\text{KBr } \nu_{\text{max}}$ : 1598  $\text{cm}^{-1}$  ( $>\text{C}=\text{O}$ , quinonemethide), 3308  $\text{cm}^{-1}$  (-OH attached to quinonemethide,  $^1\text{H}$  NMR absorption:  $\delta$  7.82, exchangeable with  $\text{D}_2\text{O}$ ) and  $^{13}\text{C}$  NMR absorption 22.5 MHz: 178.63 (dd,  $>\text{C}=\text{O}$ , quinonemethide) compare closely with the corresponding data of 15-deoxyfuerstion<sup>3</sup>, fuerstion and 3 $\beta$ -acetoxyfuerstion<sup>4</sup>. The signals at 6.98(s, 1H); 6.78(d, J=6.8Hz, 1H); 6.58(d, J=6.45Hz, 1H); 3.15 (septet, J=6.6Hz, 1H); 1.18(d, J=6.1Hz, 3H); 1.19(d, J=6.1Hz, 3H); 1.55(s, 3H); 1.49(s, 3H); 1.12(s, 3H) in the  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ , chemical shifts in  $\delta$  values with TMS as internal standard) of isobharangin compare closely with those of the protons at carbon atoms 14, 7, 6, 15, 16, 17, 18, 19 and 20 respectively of 15-deoxyfuerstion<sup>3</sup>. Thus, the structural features of rings B and C and the gem-dimethyl group of ring A of 15-deoxyfuerstion are preserved in the structure of isobharangin and this is confirmed by the proximity of corresponding chemical shifts and multiplicities of its  $^{13}\text{C}$  NMR signals with those of bharangin<sup>1</sup> and 3 $\beta$ -acetoxyfuerstion<sup>4</sup>.

The molecular formula of isobharangin results by the addition of two oxygen atoms to and subtraction of two hydrogen atoms from that of 15-deoxyfuerstion ( $\text{C}_{20}\text{H}_{26}\text{O}_2$ )<sup>3</sup>. The presence of a second hydroxy absorption at 3480  $\text{cm}^{-1}$  and another carbonyl absorption at 1715  $\text{cm}^{-1}$  in the IR spectrum of isobharangin is confirmed by the signal at 3.76 (exchangeable with  $\text{D}_2\text{O}$ ) in the  $^1\text{H}$  NMR spectrum and carbonyl signal at 209.19 (s, br)<sup>5</sup> in its  $^{13}\text{C}$  NMR spectrum suggesting that two methylene groups of ring A of 15-deoxyfuerstion are modified to a six membered ketone and a secondary hydroxy group in its structure. The contiguity of the carbonyl and hydroxyl functional groups and the location of the carbonyl group at C-2 in ring A of isobharangin is inferred from the presence of a signal at 4.14(s) assignable to the methine proton of a carbon atom carrying a secondary hydroxyl group and an AB quartet for the methylene protons adjacent to a carbonyl group (4.09, d and 2.70, d, J=14Hz) in its  $^1\text{H}$  NMR spectrum<sup>5</sup>. The hydroxyl group has been placed

unambiguously at C-3 in preference to C-1 in the structure of isobharangin (1) since, the methylene carbon atom adjacent to the carbonyl group exhibited long range  $^3J$  coupling with the protons of the methyl group at C-10 ( $18\text{-CH}_3$ ) in addition to the  $^1J$  couplings in its  $^{13}\text{C}$  NMR spectrum ( $46.16, \text{dd}, ^1J = \text{ca. } 126$  and  $\text{ca. } 141\text{Hz}$  and  $^3J = 3.7\text{Hz}$ ). The chemical shift and multiplicity



of the carbon atom carrying the secondary hydroxyl group ( $80.24, \text{d}, J = 150.1\text{Hz}$ ) and those of the remaining carbon atoms<sup>1,4</sup> at C-4 ( $47.62, \text{septet}, ^3J = 5\text{Hz}$ ); C-5 ( $158.86, \text{s, br}$ ); C-6 ( $120.93, \text{d}, ^1J = 159.9\text{Hz}$ ); C-7 ( $136.70, \text{dd}, ^1J = 162.4\text{Hz}$  and  $^3J_{\text{H-14}} = 5\text{Hz}$ ); C-8 ( $125.21, \text{s, br}$ ); C-9 ( $128.30, \text{dd}, ^3J = 2.4$  and  $6.1\text{Hz}$ ); C-10 ( $46.11, \text{s, br}$ ); C-11 ( $146.34, \text{s}$ ); C-12 ( $178.63, \text{dd}, ^3J_{\text{H-14}} = 6.7\text{Hz}, ^3J_{\text{H-15}} = 2.7\text{Hz}$ ); C-13 ( $142.49, \text{m}$ ); C-14 ( $133.29, \text{dt}, ^1J = 158.7\text{Hz}$  and  $^3J_{\text{H-7}} = ^3J_{\text{H-15}} = 6.1\text{Hz}$ ); C-15 ( $27.04, \text{doublet of septets}, ^1J = 125.6\text{Hz}, ^3J = 6.1\text{Hz}$ ); C-16 and C-17 ( $21.89, \text{qq}, J = 125.6\text{Hz}, ^3J = 5\text{Hz}$ ); C-18 ( $24.06, \text{qt}, ^1J = 129.0\text{Hz}, ^3J = 5\text{Hz}$ ); C-19 ( $27.52, \text{qq}, ^1J = 130.5\text{Hz}, ^3J = 5\text{Hz}$ ); C-20 ( $28.01, \text{qq}, ^1J = 125.6\text{Hz}, ^3J = 5\text{Hz}$ ) in the  $^{13}\text{C}$  NMR spectrum of isobharangin confirm structure (1) assigned to it. The absolute configurations of the methyl group at C-10 position (S) and the hydroxy group at C-3 position (R) in the structure of isobharangin are the same as those of  $3\beta$ -acetoxylfuerstion<sup>4</sup> in view of the similarity of their CD spectra<sup>1,4</sup>. Chair conformation for the ring A of isobharangin (1) has been inferred from a consideration of the cross signals noticed between  $18\beta\text{-CH}_3$  ( $1.55, \text{axial}$ ) and  $20\beta\text{-CH}_3$  ( $1.12, \text{a}$ );  $19\alpha\text{-CH}_3$  ( $1.49, \text{equatorial}$ ) and  $3\alpha\text{-H}$  ( $4.14, \text{a}$ );  $3\alpha\text{-H}$  and  $1\alpha\text{-H}$  ( $2.70, \text{a}$ );  $1\beta\text{-H}$  ( $4.09, \text{e}$ ) and  $18\beta\text{-CH}_3$  and  $19\alpha\text{-CH}_3$  and  $6\text{-H}$  ( $6.58$ ) in its 2D NOESY spectrum ( $\text{H}\dots\text{H}$ )<sup>6</sup>. The mass spectrum of isobharangin showed intense ions (EIMS,  $70\text{eV}$ ) at  $m/z$  (%),  $328(\text{M}^+, 16)$ ,  $257(25)$ ,  $256(93)$ ,  $229(30)$ ,  $228(100)$ ,  $213(26)$  and  $185(10)$ . Isobharangin is the first of the abietane group of diterpenoid quinonemethides with the ring A modified to a vicinal-hydroxy-keto system. Examples of similar pattern of ring A substitution system in the other class of diterpenoids have been, however, encountered earlier<sup>5,7,8,9</sup>.

The authors are grateful to Dr.G.Hohne, Technical University, Berlin for the mass spectrum, 500MHz FT-NMR National Facility, Tata Institute of Fundamental Research, Bombay and Mr.K.Kifa-tullah for the 2D and 1D NMR spectra.

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(Received in UK 30 December 1988)